INTRAMOLECULAR TRAPPING OF CYCLOBUTADIENE

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Cyclobutadiene generated from cyclobutadieneiron tricarbonyl has proven to be a useful intermediate in organic synthesis.¹ We now have evidence that cyclobutadiene will react in an intramolecular fashion with attached dienophiles. Since cyclobutadiene is a high energy, reactive molecule,² this reaction opens up new pathways to highly strained ring systems. In order to explore the possibility of this reaction, cyclobutadieneiron tricarbonyl with pendant acety-lene groups were prepared and oxidatively decomposed.

Propargyl methyl ether was photochemically added to vinylene carbonate to yield methoxymethyl-3,4-carbonyldioxycyclobutene (I) in 22% yield. This carbonyldioxycyclobutene reacted with disodium tetracarbonylferrate (-2) to produce methoxymethylcyclobutadieneiron tricarbonyl (II) in 25% yield.³

Treatment of a pentane solution of II (b.p. 50-55%/.2mm, nmr (δ) 3.4, 3.76, 4.12, 4.20, singlets, 3:2:1:2) with 48% hydrobromine acid gave a quantitative yield of bromomethyl-cyclobutadieneiron tricarbonyl III.⁴

$$CH_{3}-O-CH_{2} \longrightarrow 0 + Fe(CO)_{4} = \underbrace{Hg^{+}_{4}}_{Hg^{+}_{3}} O-CH_{2} \bigoplus Fe(CO)_{3} \xrightarrow{HBr}_{pentane} CH_{2}Br \bigoplus Fe(CO)_{3}$$

The very reactive bromomethyl moiety can serve as a link for attachment of a variety of reactive function groups. For example, potassium 3-pentyne-1-oxide reacted with III to produce IVa (n.m.r. (δ) 4.2 (s), 4.6 (s), 3.85 (s), 3.6 (t), 2.4 (m), 1.75 (t), 2:1:2:2:2:3) in a near

III +
$$KO-(CH_2)_n-C=C-CH_3$$

 $a,n = 2$
 $b,n = 1$

 $CH_3-C=C-(CH_2)_n-O-CH_2 \bigcirc Fe(CO)_3$
 IVa

2425

quantitative yield.

Direct ceric ion oxidation of IVa gave a mixture of organic products. However, photolysis of IVa in pentane gave an intermediate metal complex (V is a suggested structure) which slowly gave back IVa on standing under 40 psi of carbonmonoxide. The crude photolysis mixture gave an 83% yield of a C₁₀ (VI) organic compound on ceric ion oxidation. This C₁₀ compound (VI) was identified as 5-methyl isochroman by comparison to an authentic sample.⁵ The following is a proposed scheme for the reaction.^{6,7}



The 7-methyl isomer of VI was prepared from p-bromotoluene 5 and none of this isomer could be detected in the reaction mixture.

The 2-butyne ether complex $(IVb)^8$ gave high yields of C_g compounds on direct ceric ion oxidation. Apparently the shorter chain increased the rate of intramolecular reaction and prevented some of the intermolecular reactions observed in IVa.



The major C_g product from this reaction was 4-methylphthalan (VIII) (50% yield). The other products from this reaction are under investigation.

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- All the iron complexes showed a parent peak and the characteristic consecutive loss of 28 m/e.